

It has been found that interference for free rotation can be minimised by

- (1) Ring Bending ortho H & ortho substituents
- (2) Compressing C-H (ortho bonds)
- (3) Stretching intramolecular bonds
- (4) Deforming the angle between benzene and interannular bonds.
- (5) Deforming benzene rings themselves.

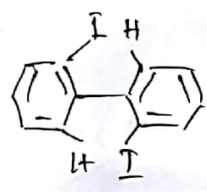
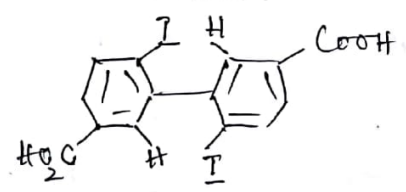
Order of - activation energy for racemisation

I > Br > Cl > NO₂ > COOH > OMe > F > H.

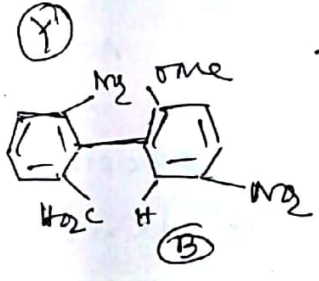
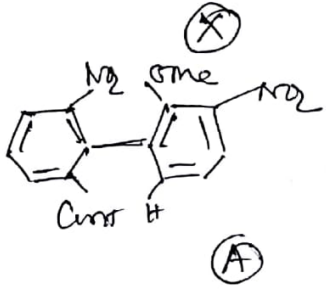
Buttressing Effect -

Found that - the activation energy of racemisation for active biphenyls is ~~increased~~ increased when a substituent is present in meta position to a bond joining the rings. This is called buttressing effect.

Reason: Prevention of outward bending of ortho substituents.



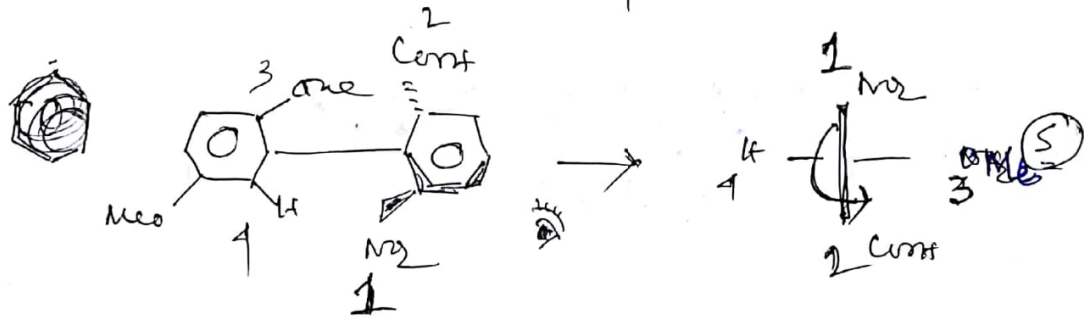
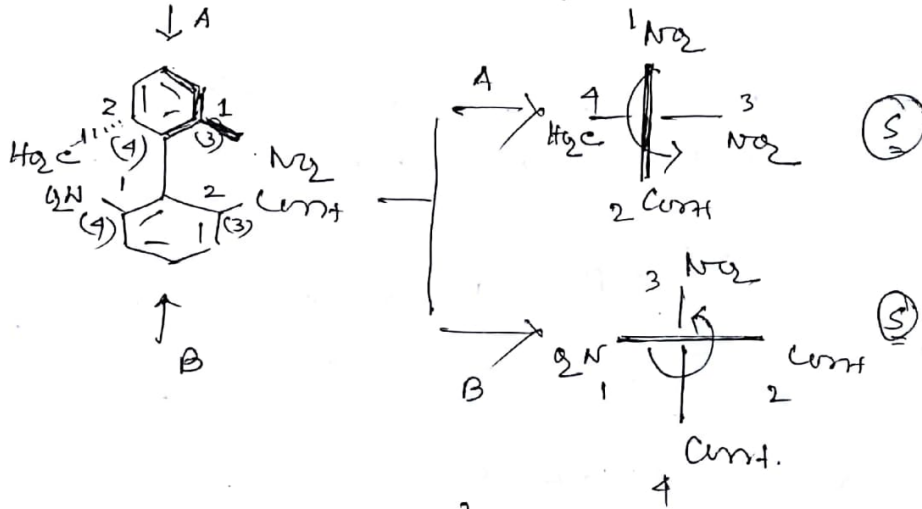
* $E_{act}(X) - E_{act}(Y) \approx 21 \text{ kJ/mol}$



* (B) racemises faster than (A)

Nomenclature

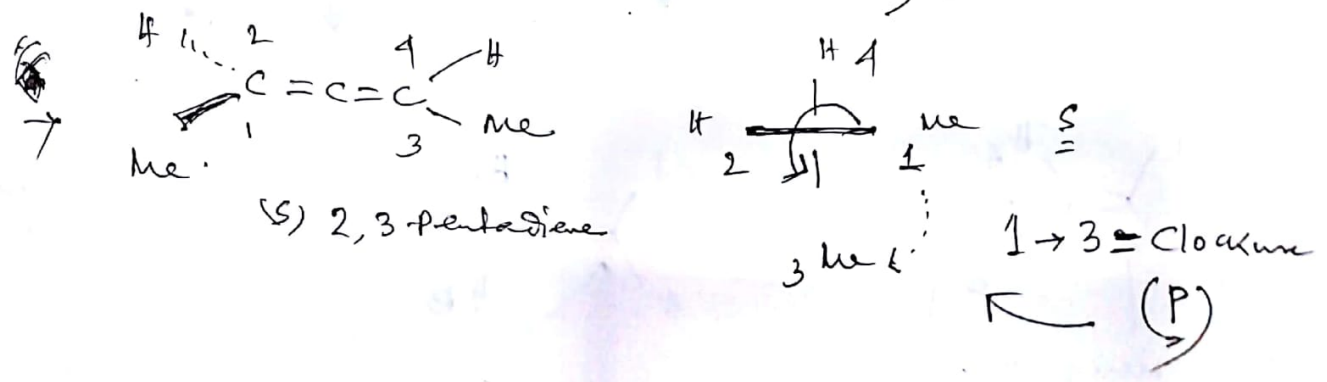
Same as alkenes.



P, M descriptors for molecules with Chiral axis

For assigning helical descriptors, only the groups
 1) - highest priority nearest to the view direction and
 the far are considered (1 → 3).

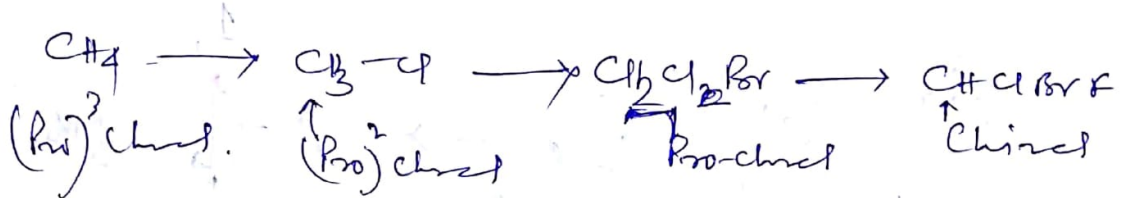
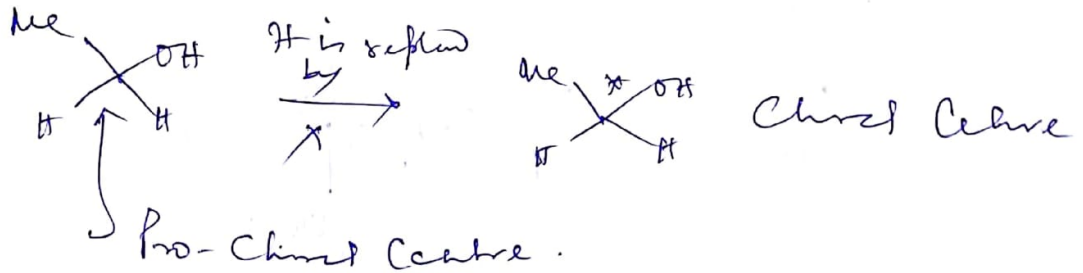
If 1 → 3 clockwise it is P (plus) or if
 anti clockwise it is M (minus).



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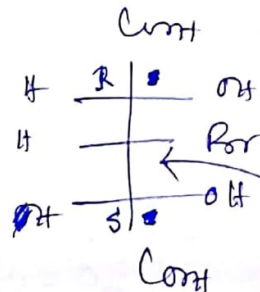
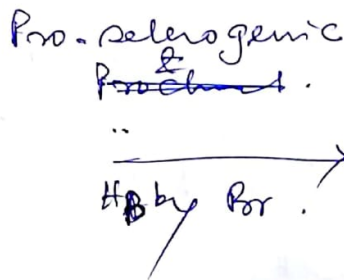
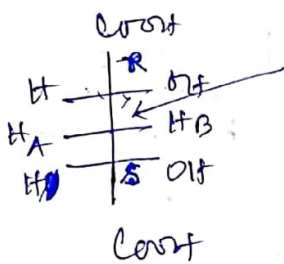
Pro-chirality

Replacement of one ligand leading to chiral centre, known as pro-chiral centre.

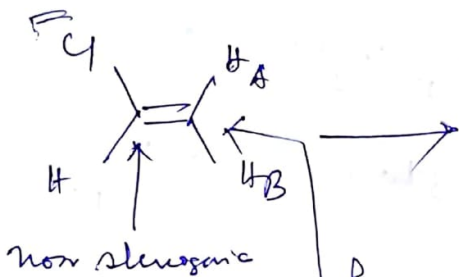


Pro-stereogenic centres

Replacement of centre bearing homomorphic ligands or groups which can be converted into a stereogenic centre by replacing one of them by different substituents.

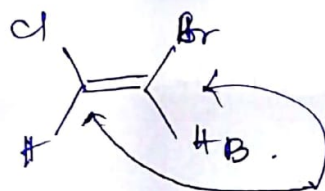


Stereogenic \neq ~~Chiral~~ achiral



non stereogenic

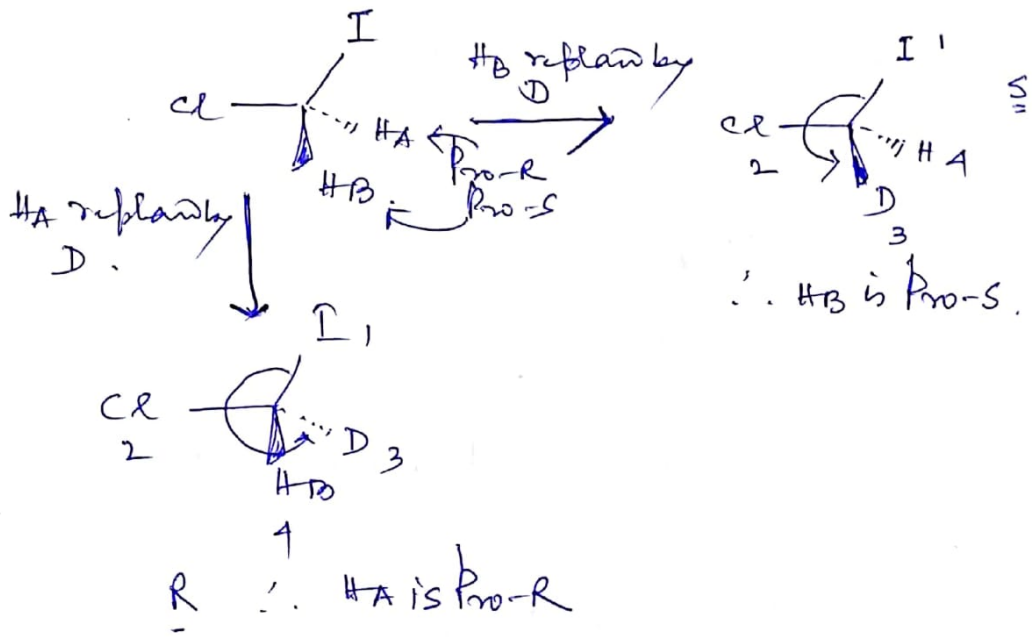
Pro-stereogenic but not pro-chiral.



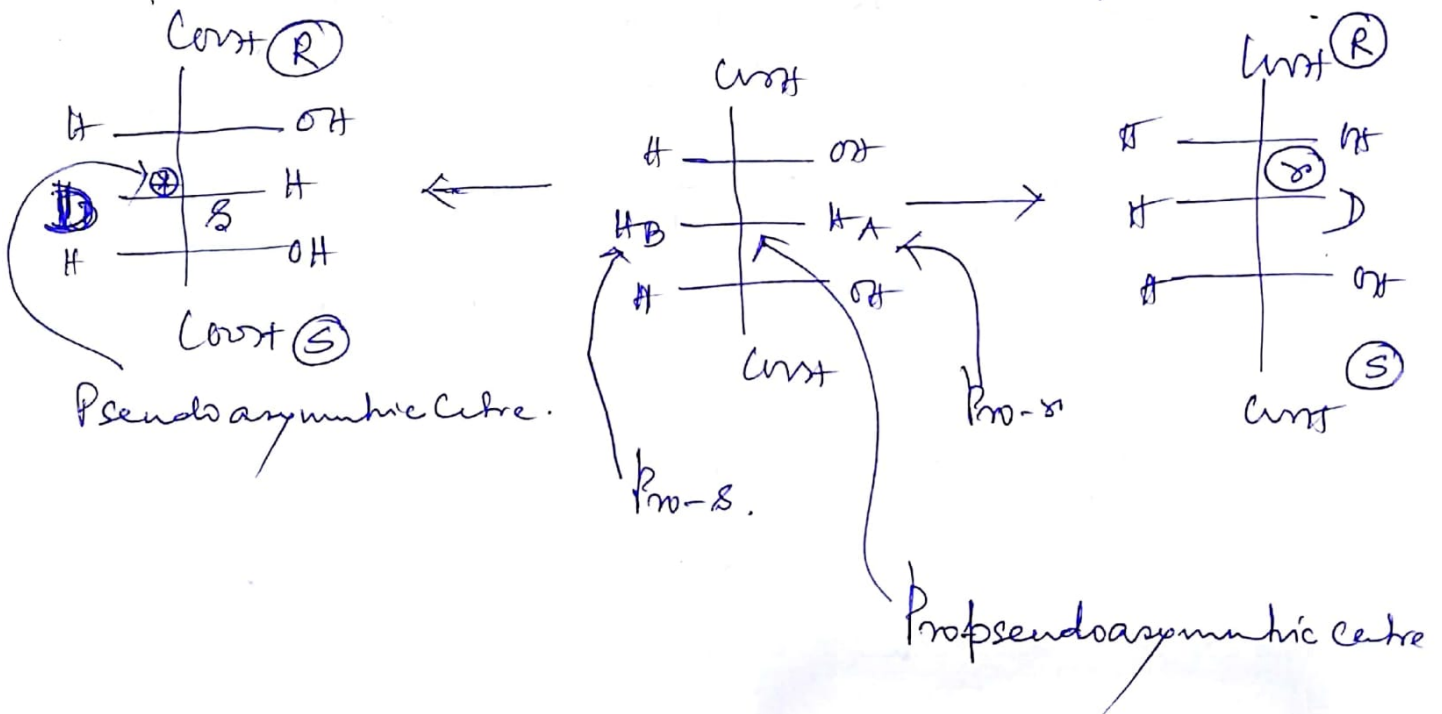
Stereogenic Centre but not-chiral.

Pro-R & Pro-S.

(11) (12)

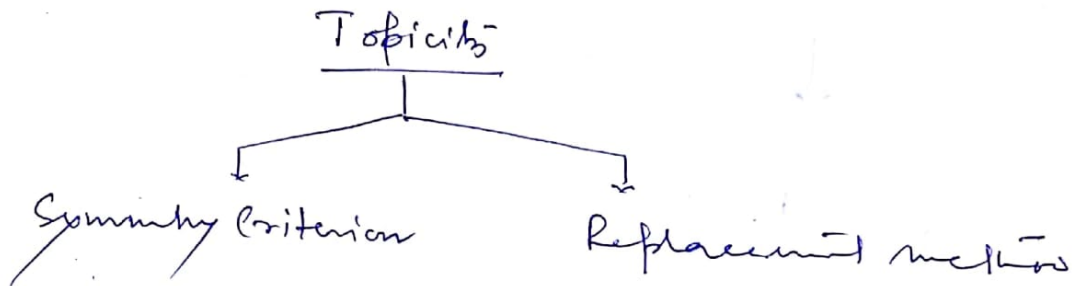


In case of Pro-pseudoasymmetric centre the ligands are named Pro-r and Pro-s.



Topics of ligands.

Nature of two homomorphic ligands (atoms or groups).

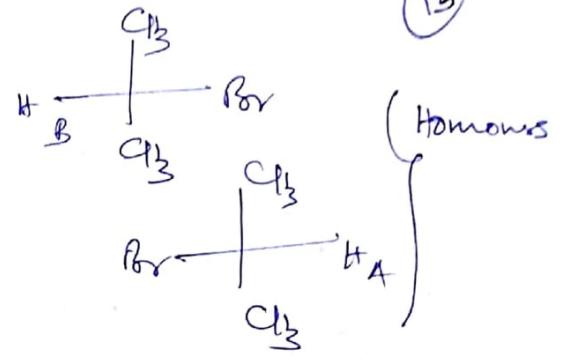
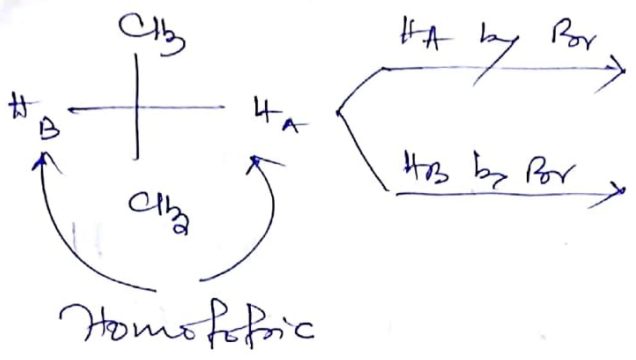


* Symmetry method (Criterion)

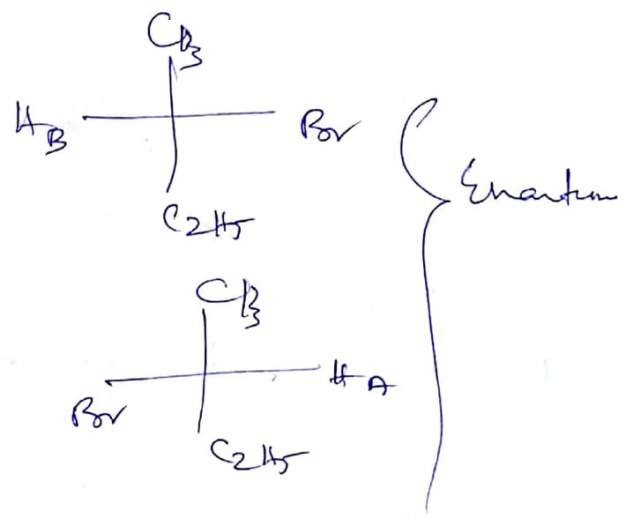
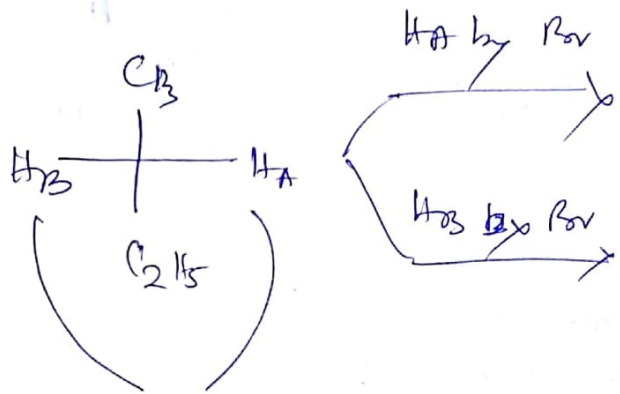
1. Replaced by C_2 or exchangeable by C_2 - Homotopic
2. Exchanged by σ - enantiotopic
3. Not exchangeable w.r.t C_2 or σ → Diastereotopic
4. Chemical environments are different
↓
Constitutionally heterotopic

* Replacement method

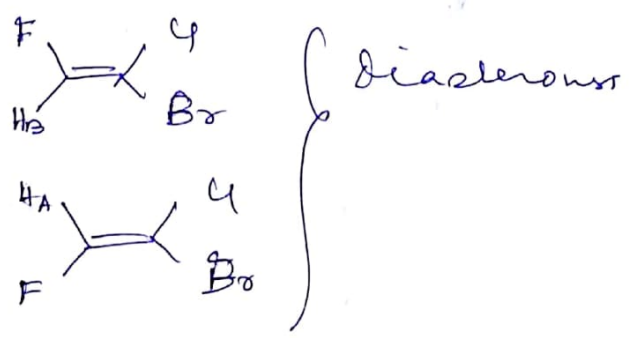
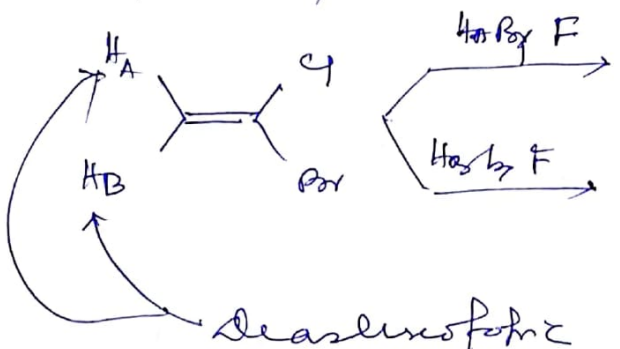
- | | | | |
|------------------------|----------|---|--|
| Replacement leading to | homomers | → | Homotopic |
| " | " | " | Enantiotopic |
| " | " | " | Diastereotopic |



Exchangable with C_2 .



Exchangable with σ



non exchangeable with C_2 or σ .

